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by the sharing and trading of electrons between atoms. Understanding how electrons are arranged in an atom is a building block of Chem I. Electrons in an atom are contained in specific energy levels (1, 2, 3, and so on) that are different distances from the nucleus.

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convert kilojoules to joules, divide the
number of kilojoules by lost as heat. 12.
When a fuel is burned, some of its

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chemical potential energy is 11. One calorie equals 4.184 joules.

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